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New α -Functionalized Phosphorus Acid Surfactants: Synthesis, Dissociation Constants and Molecular Aggregation

D. Albouy^a; A. Brun^a; G. Etemad-moghadam^a; A. Munoz^a; I. Rico-lattes^a Laboratoire des IMRCP (UMR 5623) Université Paul Sabatier, Toulouse cedex 4, FRANCE

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New α-Functionalized Phosphorus Acid Surfactants: Synthesis, Dissociation Constants and Molecular Aggregation

D. ALBOUY, A. BRUN, G. ETEMAD-MOGHADAM*, A. MUNOZ and I. RICO-LATTES

Laboratoire des IMRCP (UMR 5623) Université Paul Sabatier – 118, route de Narbonne, 31062 Toulouse cedex 4 -FRANCE.

Pursuing our investigation into the reactivity of red phosphorus, its hydrolysis derivatives (PH₃ and H₃PO₂), ^{1,2} and P-H labile compounds, ³ we report here a convenient method for the direct synthesis of free α-hydroxyalkyl phosphinic acid surfactants from aqueous hypophosphorous acid and the long-chain aldehydes or imines under sonication.

The acids 1-3 can be considered as relatively strong acids and their extractant properties are under investigation. Self-association behavior of their ammonium salts shows the formation of micelles for the single-chain amphiphiles 1-2 and of vesicles for the bolaform 3. The dissociation constants and the critical micellar concentrations (CMC) are determined by potentiometry, tensiometry and ³¹P NMR.

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^{*} Tel: 05 61 55 83 33 - E-mail: etemad@ramses.ups-tlse.fr